# We produce alloyed nanoparticles by reducing doped perovskites.

## Tailoring bi- and trimetallic nanoparticles by exsolution from perovskites

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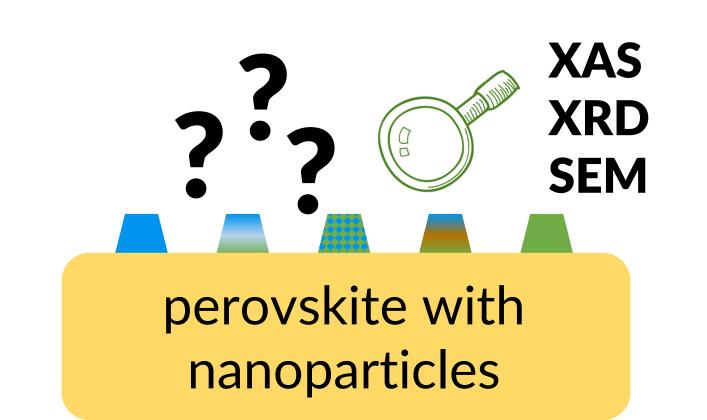
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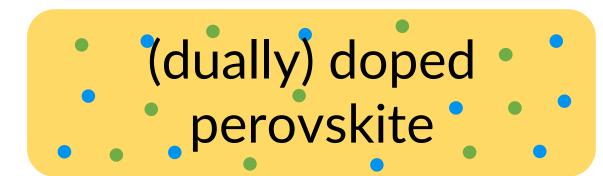
#### Introduction

Reducing a doped perovskite oxide can trigger **exsolution**, when the dopants emerge on the surface forming metal **nanoparticles**. In the case of multiple doping, this process is rather complex, various behaviors and thus resulting particle compositions are conceivable.

Focusing on the critical parameter of the **temperature** during reduction, this study investigates several doped variants of the perovskite  $Nd_{0.6}Ca_{0.4}FeO_{3-\delta}$  (NCF), in order to understand the system dually doped with **Co** and **Cu**.



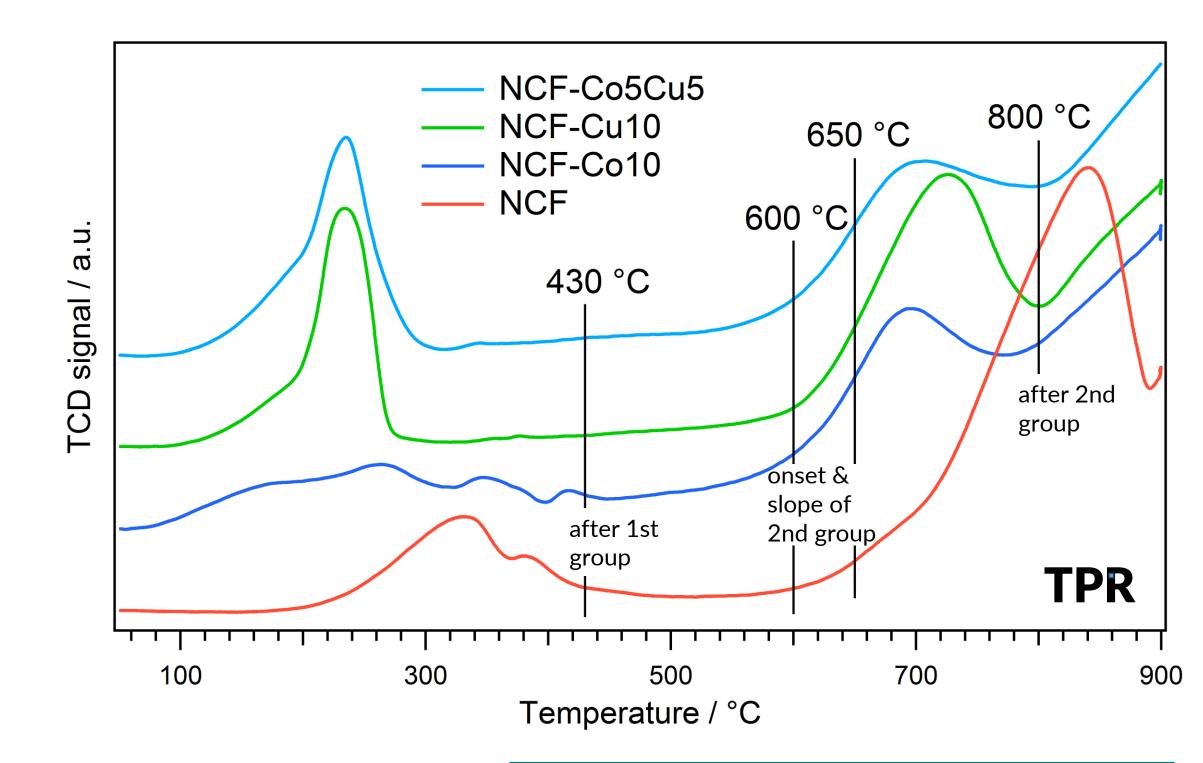




10 % Co: NCF-Co10 10 % Cu: NCF-Cu10 5 % Co + 5 % Cu: NCF-Co5Cu5

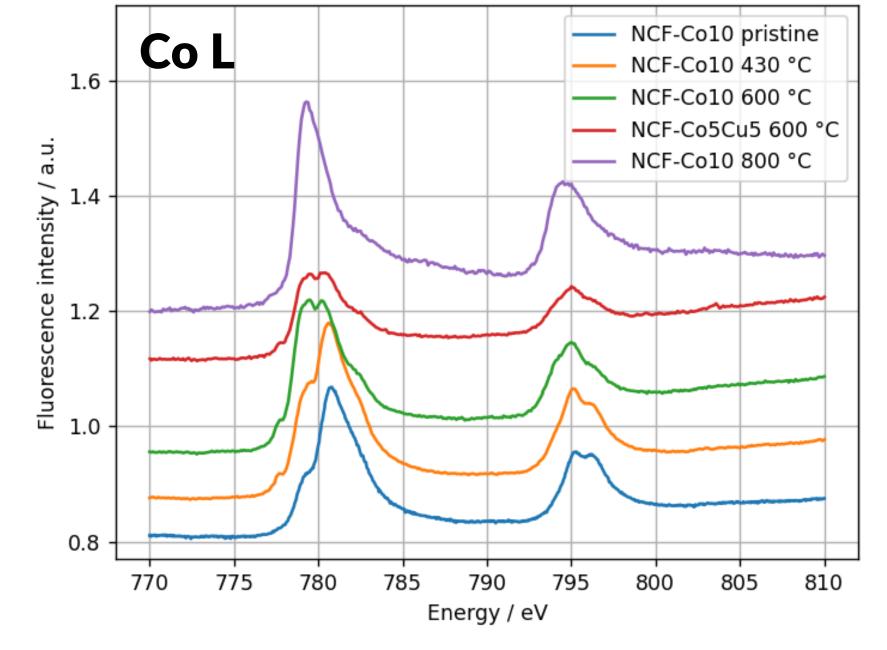
#### Approach

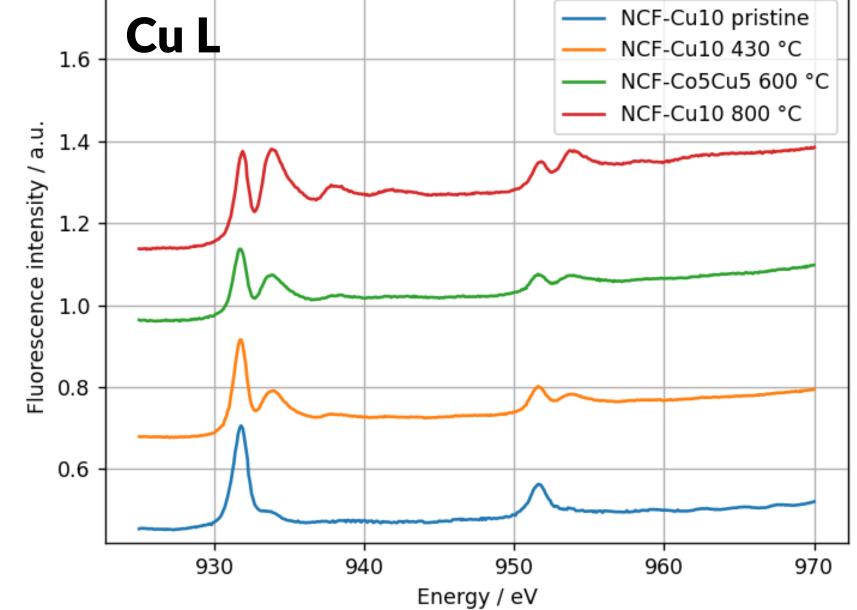
The materials have been reduced in 5 % H<sub>2</sub> at different temperatures, which were chosen according to interesting points in the profiles of temperature programmed reduction (TPR) experiments. Afterwards, they were investigated using X-ray diffraction (XRD), X-ray absorption (XAS), and Scanning Electron Microscopy (SEM).



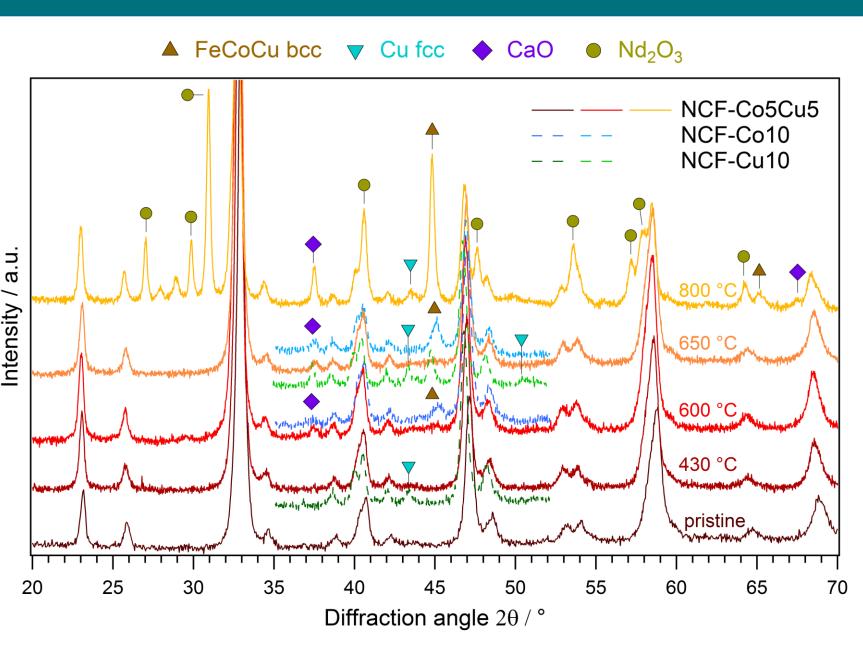
Comparison of the spectra shows both metallic Co and Cu species above 600 °C in NCF-Co5Cu5.

### Results XAS



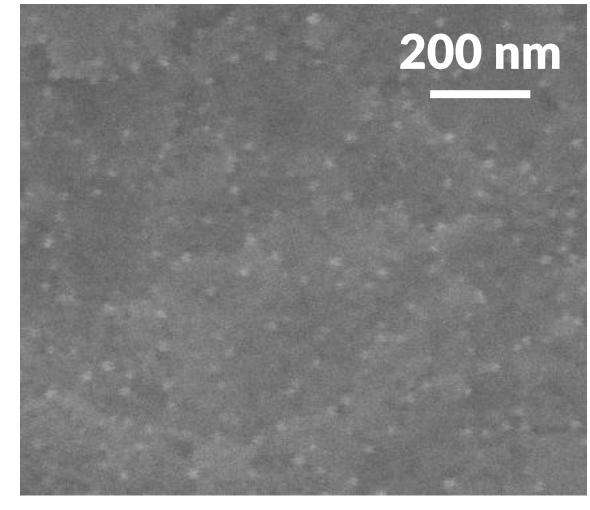


#### **Results SEM**

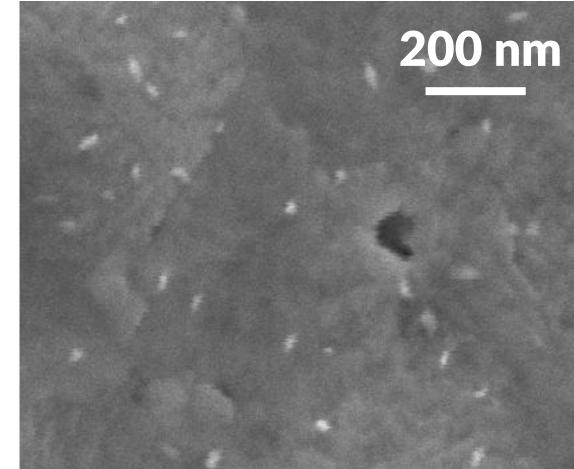


Results XRD

In NCF-Co5Cu5, only a metallic bcc phase is seen.



NCF-Co10 after 600 °C: spheres, size ~30 nm.



NCF-Co5Cu5 after 600 °C: needle-like, length ~40 nm.

#### Conclusion



- Exsolution starts at ~600 °C.
- The particles are **alloyed**, consisting of both Co and Cu, as well as the host element Fe, resulting in a metallic **bcc phase**.
- These particles are needlelike with length ~40 nm.
- Above ~700 °C, the material decomposes.









Take a picture to find out how exsolution can be used for catalysis.



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